

上海交通大学

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300-500

[Red callout box]

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HCCI

HCCI

NO_x

HCCI

DME
Lawrence Livermore

DME
DME

DME
HCT

EGR

HCCI

DME HCCI

H₂O₂

H₂ CO

EGR CH₄ CH₃OH

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EGR

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NUMERICAL SIMULATION OF HOMOGENEOUS CHARGE COMPRESSION IGNITION COMBUSTION FUELED WITH DIMETHYL ETHER

Times New
Roman

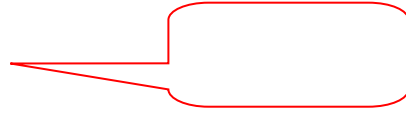
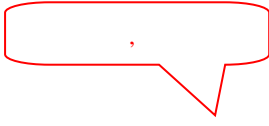
ABSTRACT

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HCCI (Homogenous Charge Compression Ignition) combustion has advantages in terms of efficiency and reduced emission. HCCI combustion can not only ensure both the high economic and dynamic quality of the engine, but also efficiently reduce the NO_x and smoke emission. Moreover, one of the remarkable characteristics of HCCI combustion is that the ignition and combustion process are controlled by the chemical kinetics, so the HCCI ignition time can vary significantly with the changes of engine configuration parameters and operating conditions. In this work numerical scheme for the ignition and combustion process of DME homogeneous charge compression ignition is studied. The detailed reaction mechanism of DME proposed by American Lawrence Livermore National Laboratory (LLNL) and the HCT chemical kinetics code developed by LLNL are used to investigate the ignition and combustion processes of an HCCI engine fueled with DME. The new kinetic mechanism for DME consists of 79 species and 399 reactions. To consider the effect of wall heat transfer, a wall heat transfer model is added into the HCT code. By this method, the effects of the compression ratio, the fuel-air equivalence ratio, the intake charge heating, the engine speed, EGR and fuel additive on the HCCI ignition and combustion are studied. The results show that the HCCI combustion fueled with DME consists of a low temperature reaction heat release period and a high temperature reaction heat release period. It is also founded that increasing the compression ration, the equivalence ratio, the intake charge temperature and the content of H_2O_2 , H_2 or CO cause advanced ignition timing. Increasing the engine speed, adoption of cold EGR and the content of CH_4 or CH_3OH will delay the ignition timing.

Key words: HCCI, chemical kinetics, numerical simulation, DME, EGR



.....	1
1.1 HCCI	1
1.1.1 HCCI	1
.....	
1.4	1
DME	2
2.1	2
2.1.1	2
.....	
.....	
.....	4
.....	5
.....	6

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NO_x PM
 NO_x+HC <2.5g/bph-hr PM <0.05g/bph-hr

[1]

(NO_x)

(CO),

IV

NO_x

(PM)

NO_x

trade-off

2

HCCI

HCCI

HCCI

HCCI

HCCI

[2]

2

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2

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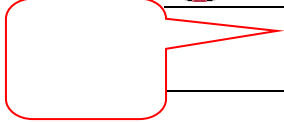
$$m = \sum_{k=1}^K m_k$$

2-1

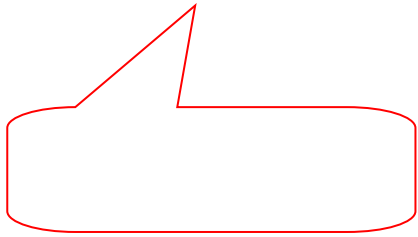
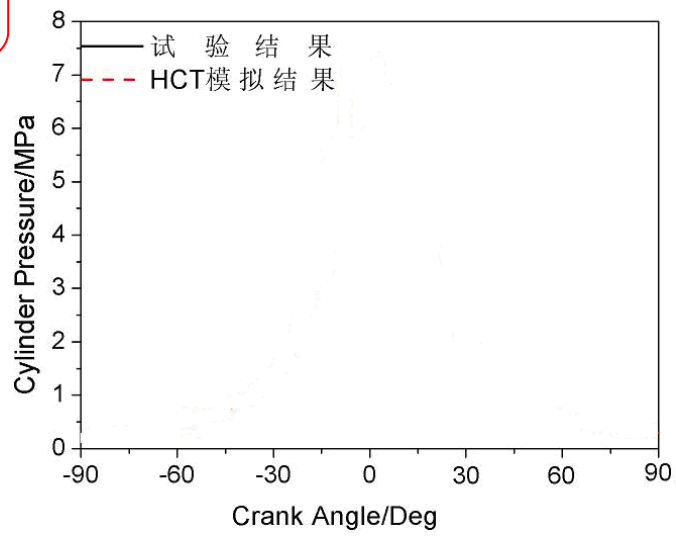
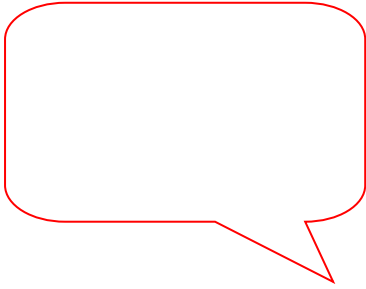
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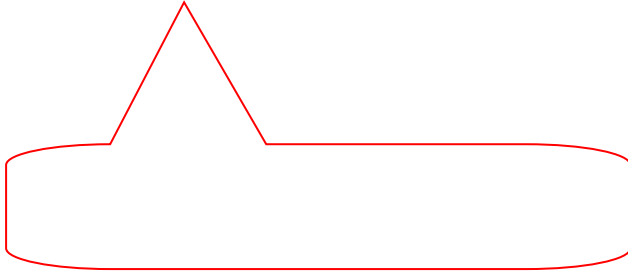
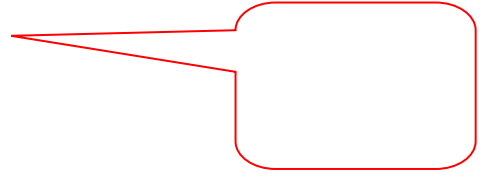
$$f(x,y) = f(0,0) + \frac{1}{1!} \left(x \frac{\partial}{\partial x} + y \frac{\partial}{\partial y} \right) f(0,0) + \frac{1}{2!} \left(x \frac{\partial}{\partial x} + y \frac{\partial}{\partial y} \right)^2 f(0,0) + K + \frac{1}{n!} \left(x \frac{\partial}{\partial x} + y \frac{\partial}{\partial y} \right)^n f(0,0) + K$$

	H _f (kcal/mol)	S _f (kcal/mol)	C _p (kcal/mol)
A1			
A2			
A3			

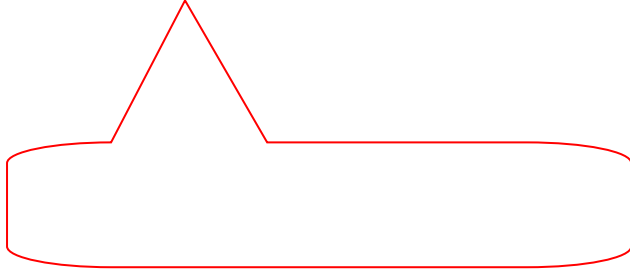
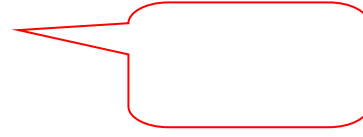


	$H_f(\text{kcal/mol})$	$S_f(\text{kcal/mol})$	$C_p(\text{kcal/mol})$
A4			
A5			
A6			
A7			
A8			





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